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Enzyme conversion of mulberry red pigments in a microfluidic aqueous two-phase system

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CONTNET

Introduction



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Acknowledgments



01 Introduction

Mulberry red pigment--natural edible pigment



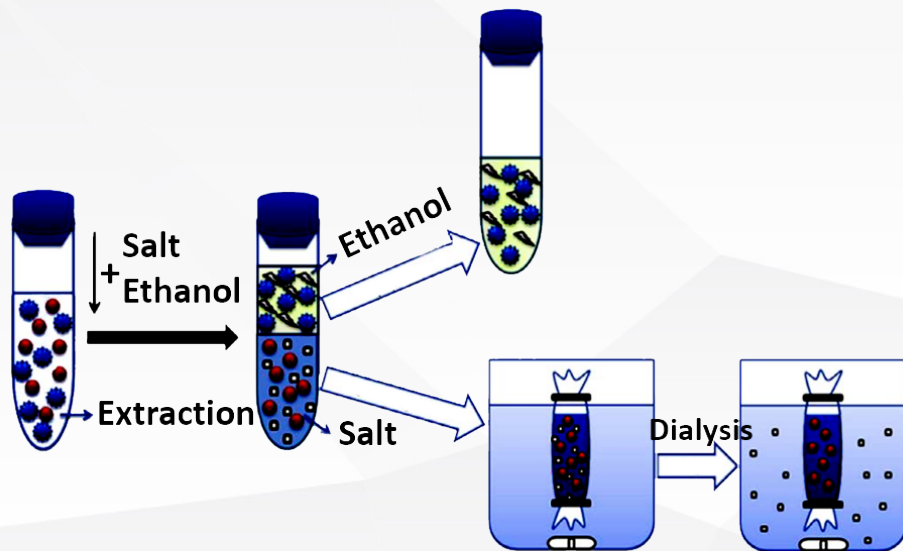
Aqueous two-phase enzyme catalysis system

Extraction, catalysis, separation

Disadvantages:

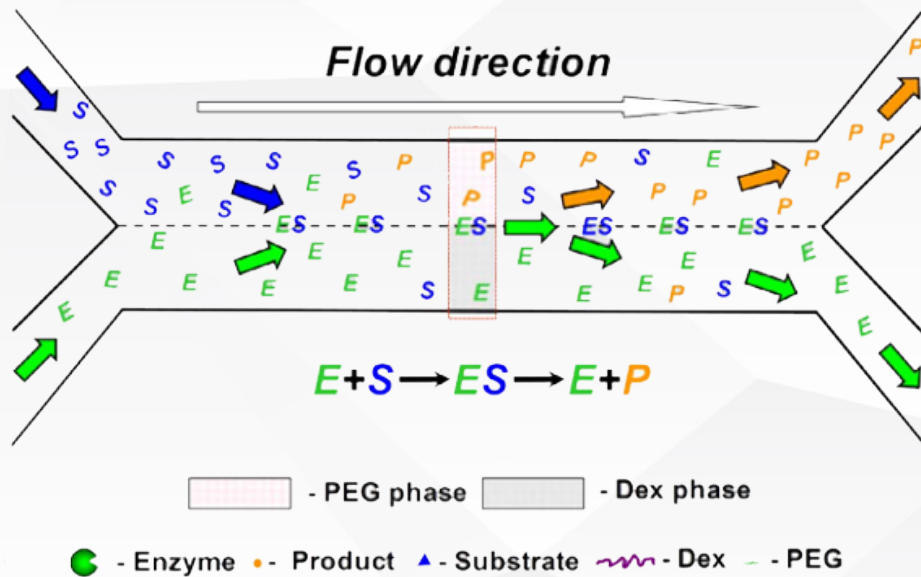
Long separation time
Easily emulsification

Microfluidic

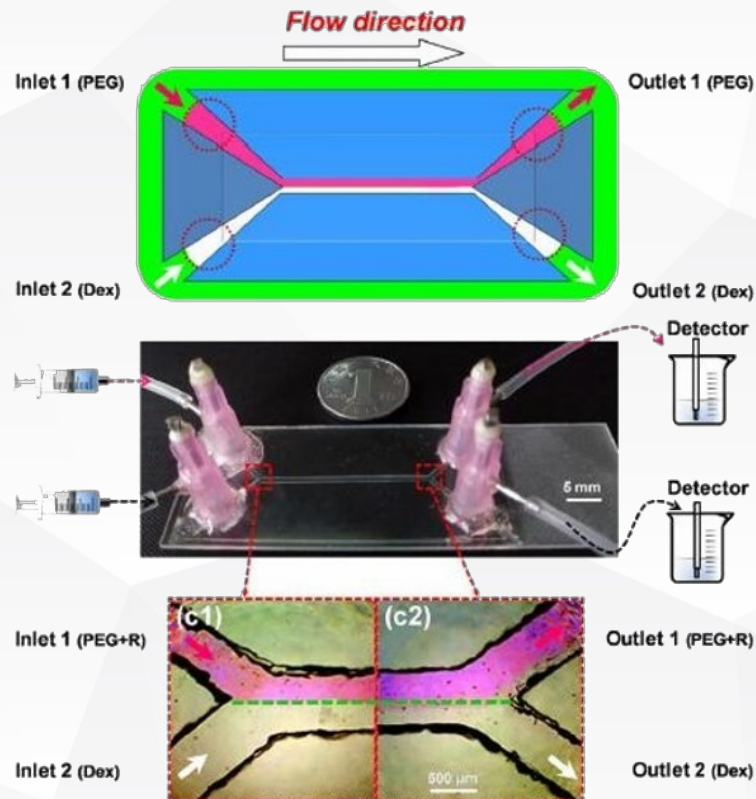


● Polysaccharide ● Protein □ Salt ◂ Ethanol

Microfluidic aqueous two-phase enzyme catalysis system



500 times higher than enzymatic rate in conventional reactor



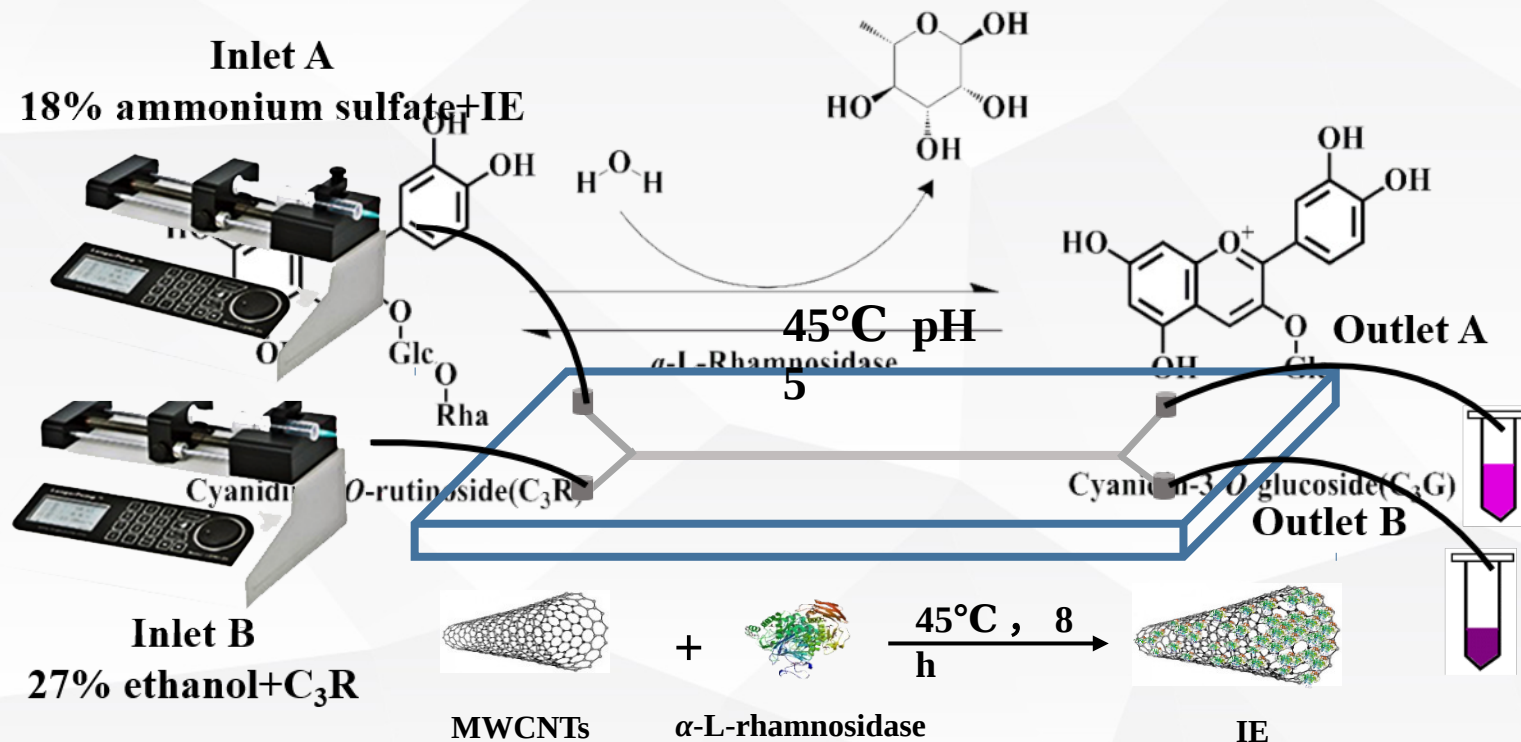


Fig. 1 Diagram of microfluidic aqueous two-phase device.

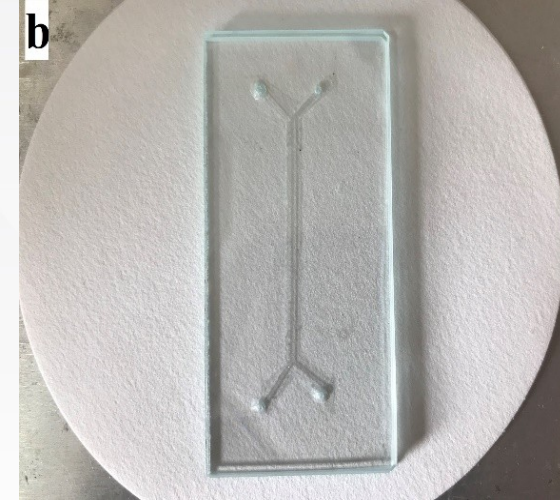
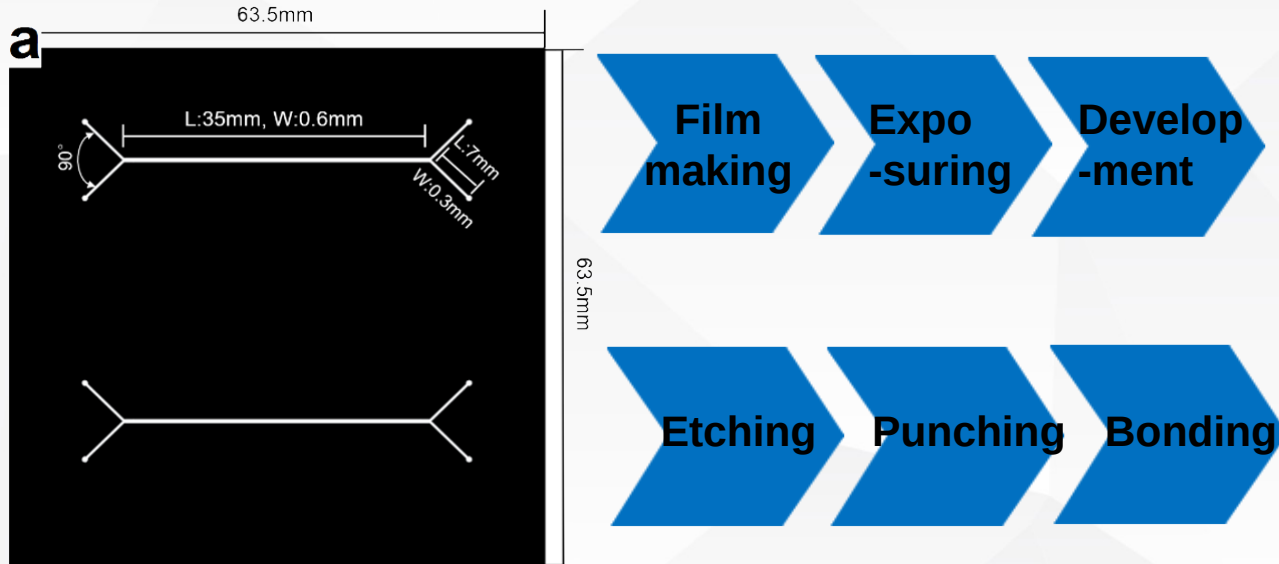


Fig.2 Design and manufacture of double Y-branched microfluidic chip.

(a) Design figure of double Y-type microfluidic chip; (b) Real figure of double Y-branched microfluidic chip.

The middle main channel is 3.5 cm length and 600 μm width
The double Y-branch is 7 mm length and 300 μm width

Inlet A →
Ammonium sulfate

Longer pump

Inlet B →
Ethanol

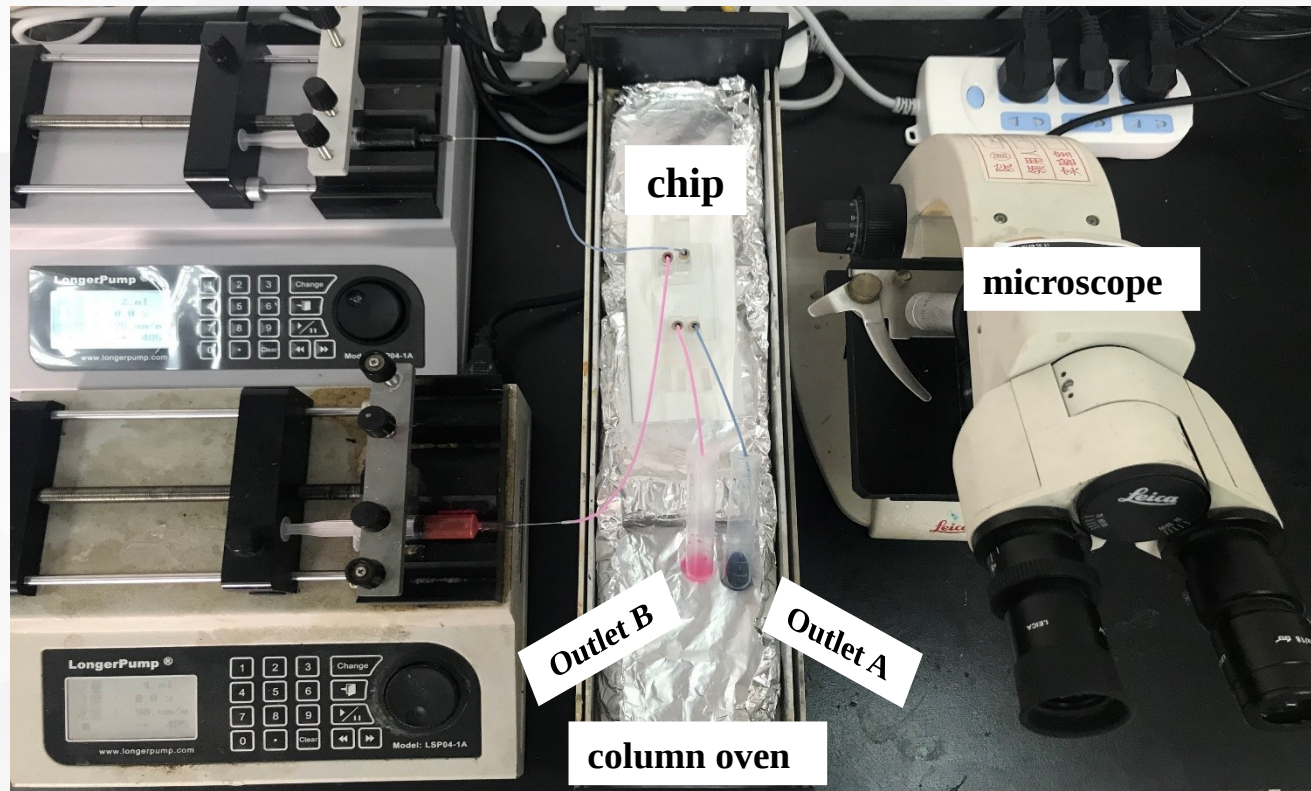
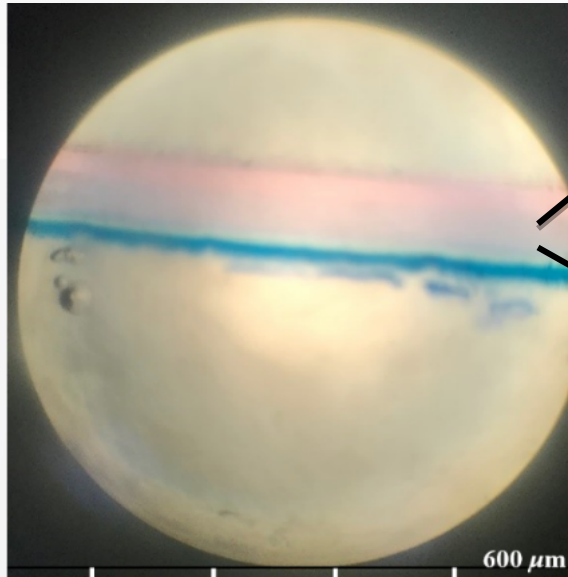


Fig. 3 Physical diagram of microfluidic aqueous two-phase device.



18% Ethanol and Rhodamine B
11.5 $\mu\text{L}/\text{min}$

27% Ammonium sulfate and
malachite green
15.5 $\mu\text{L}/\text{min}$

$$Re = \frac{dvp}{\mu}$$

$$Re < 2000$$

parallel flow formation

Tab. 1 Calculation of Re in microchannel.

| Substance | Width (μm) | Flow rate ($\mu\text{L}/\text{min}$) | Density (kg/m^3) | Viscosity ($\text{Pa}\cdot\text{s}$) | Re |
|------------------|----------------------------|---|---------------------------------------|---|-----------------------|
| Ethanol | 900 | 11.5 | 7.89×10^2 | 1.074×10^{-3} | 1.11×10^{-3} |
| Ammonium sulfate | 900 | 15.5 | 1.42×10^3 | 2×10^{-3} | 1.48×10^{-3} |

Fig. 4 Photograph of ATPS in microchannel under microscope.

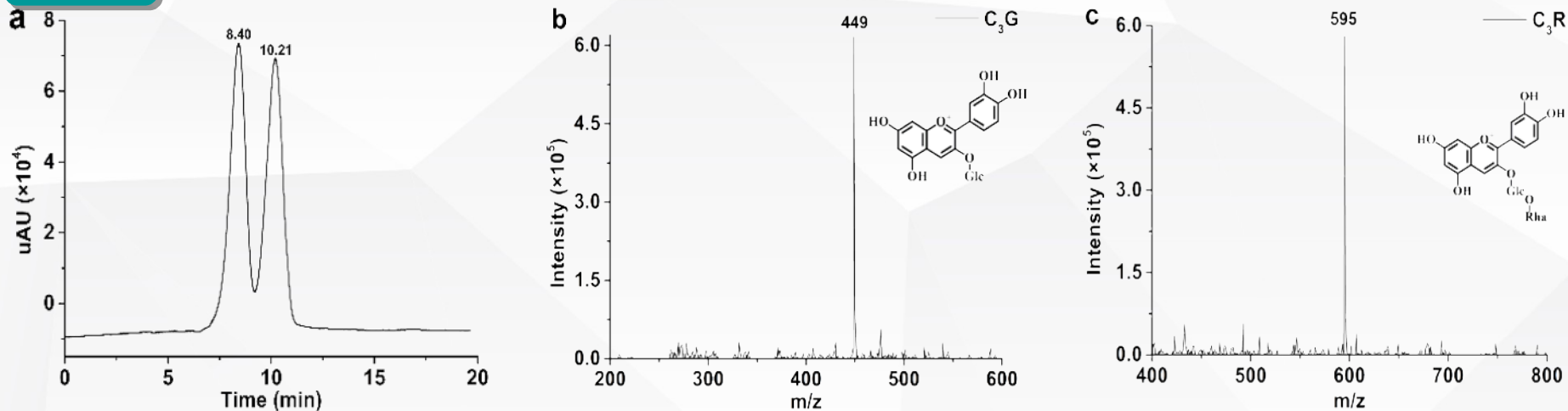
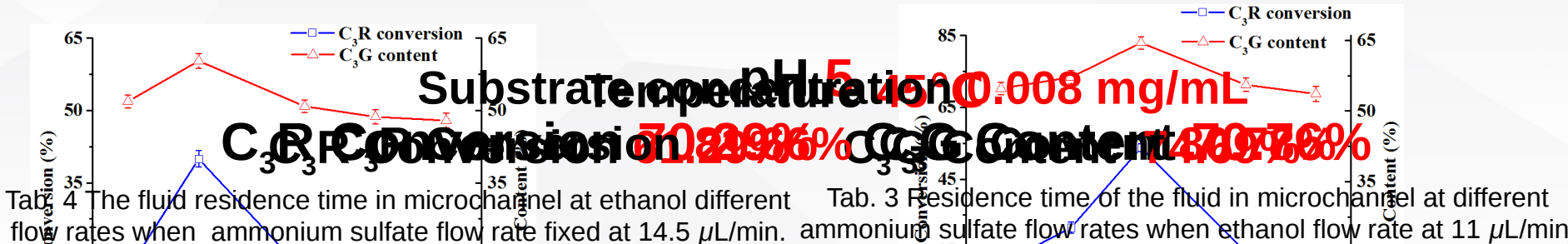


Fig. 5 HPLC–PDA-MS/MS of the product in microfluidic aqueous two-phase enzyme catalytic system. (a) HPLC chromatograms of mulberry red pigment; (b) Mass chromatogram of C_3G ; (c) Mass chromatogram of C_3R .

Tab. 2 HPLC-PDA-ESI-MS data of mulberry red pigment.

| Peak | Rt (min) | $[\text{M}]^+$ (m/z) | λ_{max} (nm) | Content |
|------|----------|--------------------------|-----------------------------|----------------------|
| 1 | 8.40 | 449 | 280,513 | C_3G |
| 2 | 10.21 | 595 | 280,513 | C_3R |



| Ethanol flow rate (μL/min) | Ammonium sulfate flow rate (μL/min) | Residence time τ (s) | Ethanol flow rate (μL/min) | Ammonium sulfate flow rate (μL/min) | Residence time τ (s) |
|----------------------------|-------------------------------------|----------------------|----------------------------|-------------------------------------|----------------------|
| 8.00 | 14.50 | 10.50 | 11.00 | 13.50 | 10.50 |
| 8.50 | 14.50 | 10.12 | 11.00 | 14.00 | 10.12 |
| 9.00 | 14.50 | 9.77 | 11.00 | 14.50 | 9.77 |
| 10.00 | 14.50 | 8.59 | 11.00 | 16.00 | 8.86 |
| 11.00 | 14.50 | 8.33 | 11.00 | 16.50 | 8.59 |
| 11.50 | 14.50 | 8.10 | 11.00 | 17.00 | 8.33 |
| 12.00 | 14.50 | 7.88 | 11.00 | 17.50 | 8.10 |
| 12.50 | 14.50 | 7.46 | 11.00 | 18.00 | 7.88 |

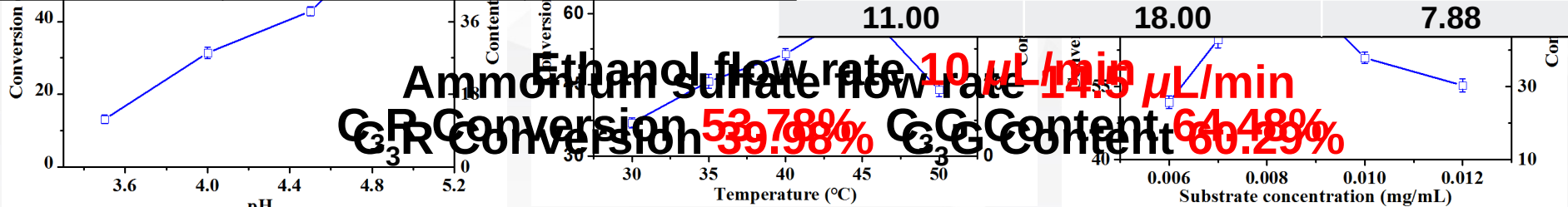


Fig. 6 Effect of conditionsons on C₃R conversion rate and C₃G content.

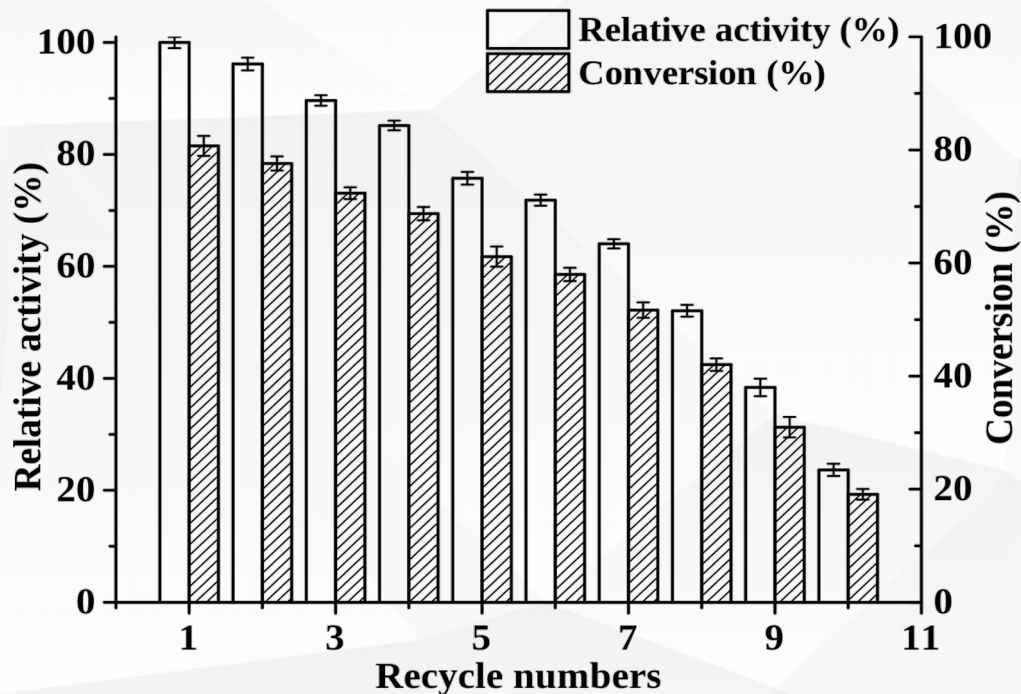
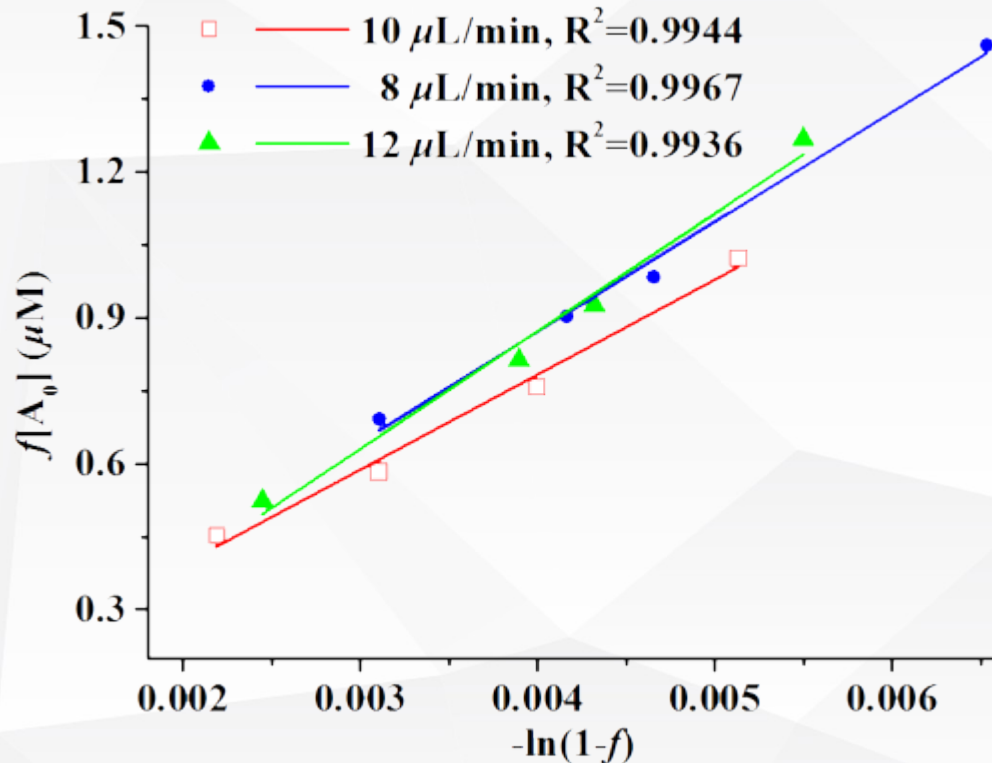


Fig. 7 Reuse of immobilized enzyme.

Immobilized enzyme can be reused **7** times
Relative enzyme activity is higher than **60%**



$$f[A_0] = \frac{C}{Q} + K_m \ln(1-f)$$

Tab. 5 Different K_m of ethanol flow rates when ammonium sulfate flow rate fixed at 14.5 $\mu\text{L}/\text{min}$

| Ethanol flow rate ($\mu\text{L}/\text{min}$) | K_m (μM) |
|--|-------------------------|
| 12 | 242.53 |
| 8 | 226.14 |
| 10 | 195.39 |

Fig. 8 Lilly-Hornby plots for immobilized enzyme in photopatterned microchannel.

- 1 The size of **double Y-branched** microfluidic chips: middle main channel: 35 mm×0.6 mm × 0.15 mm, double Y-branch: 7 mm×0.3 mm×0.15 mm.
- 2 Under the optimal conditions of ammonium sulfate flow rate of **14.50 $\mu\text{L}/\text{min}$** , ethanol flow rate of **10 $\mu\text{L}/\text{min}$** , pH **5**, temperature **45°C** and substrate concentration of **0.008 mg/mL**, the conversion of C₃R and the content of C₃G were **82%** and **81%**, respectively.
- 3 The immobilized enzyme could be reused **7** times, the relative enzyme activity was stabilized at more than 60% and the C₃R conversion rate was maintained at more than 50%.
- 4 When the ethanol flow rate was **10 $\mu\text{L}/\text{min}$** , the K_m value was the lowest and the enzyme and substrate concentration had the highest affinity.

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Thanks for listening